

STRUCTURES OF ISOMBAMICALCONE AND LOPHIROCHALCONE,
BI- AND TETRA-FLAVONOIDS FROM *LOPHIRA LANCEOLATA*

R. GHOGOMU TIH^a, B.L. SONDEGAM^a, M.T. MARTIN^b and B.BODO^b

^aDépartement de Chimie organique, Université de Yaoundé, B.P. 812, Yaoundé, Cameroun.

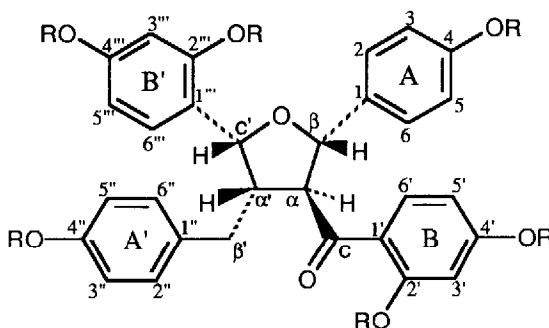
^bLaboratoire de Chimie, UA 401, CNRS, Muséum national d'Histoire naturelle,

63, rue Buffon, 75231 Paris Cedex 05, France.

Summary: From the stem bark of *Lophira lanceolata* two new flavonoids, a chalcone dimer, isombamichalcone **1**, and a tetraflavonoid, lophirochalcone **4** have been isolated.

The stem bark of the Cameroonian medicinal plant, *Lophira lanceolata* (Ochnaceae) has been shown to be rich in biflavonoids with marked structural particularities [1-3]. From the polar fraction of the AcOEt extract, we have isolated isombamichalcone, and a tetraflavonoid lophirochalcone, for which structures **1** and **4** respectively have been assigned from spectroscopic and chemical evidences. Both compounds were obtained after purification by repeated column chromatography on silica gel with eluent mixture (1-CH₂Cl₂/AcOEt gradient; 2-CH₂Cl₂/MeOH, 10:1), and showed positive flavonoid test.

- 1**: R = H
2: R = CO CH₃
3: R = CH₃



Isombamichalcone, **1** (C₃₀H₂₆O₈), amorphous colorless powder, $[\alpha]_D^{20} = -129^\circ$ (c 0.5, Me₂CO) gave $[M+H]^+$ ion at m/z 515 in CIMS, confirming a MW of 514 and showed IR absorption bands (KBr disc) $\nu_{cm^{-1}}$: 3345 (OH), 1625 (conjugated and chelated C=O), 1618, 1515 (aromatic ring). Complete acetylation (Ac₂O/Py) of **1** gave an amorphous hexa-acetate **2** (C₄₂H₃₈O₁₄, M:766), whose ¹H NMR spectrum (Me₂CO-d₆) had six sharp singlet signals (each 3H) at 2.293, 2.273, 2.269, 2.260, 2.217 and 2.153 ppm, indicative of six OH groups in **1**. Permethylation of **1** with CH₃I/K₂CO₃ in dry acetone afforded a hexa-

TABLE 1. ^{13}C and ^1H NMR data for isombamichalcone **1** and lophirochalcone **4**.

C	1		4, [I]		4, [II]	
	δ_{C} (m)	δ_{H} (m) J	δ_{C} (m)	δ_{H} (m) J	δ_{C} (m)	δ_{H} (m) J
1'	113.5 s		116.3 s		110.3 s	
2'	166.6 s		*167.5 s		*166.8 s	
3'	103.3 d	6.177 d 2.4	103.8 d	6.351 d 2.4	103.4 d	6.186 d 2.4
4'	165.5 s		*165.6 s		*165.7 s	
5'	108.3 d	5.909 dd 8.9;2.4	109.2 d	6.560 dd 9.0;2.4	108.5 d	5.958 dd 9.1;2.4
6'	133.7 d	6.444 d 8.9	133.2 d	7.920 d 9.0	133.8 d	6.410 d 9.1
C	205.5 s		192.6 s		205.5 s	
α	59.6 d	3.796 dd 6.5;3.7	117.9 d	7.284 d 15.2	60.1 d	3.702 dd 6.5;3.5
β	84.0 d	5.169 d 6.5	145.6 d	7.684 d 15.2	84.4 d	5.138 d 6.5
1	133.3 s		128.3 s		131.8 s	
2	128.2 d	7.310 m	126.4 d	7.011 d 1.8	128.3 d	7.166 m
3	116.2 d	6.841 m	132.7 s		116.3 d	6.837 m
4	157.7 s		*157.0 s		°157.9 s	
5	116.2 d	6.841 m	110.3 d	6.844 d 6.8	116.3 d	6.837 m
6	128.2 d	7.310 m	132.4 d	7.512 d 6.8;1.8	128.3 d	7.166 m
1'''	117.3 s		121.3 s		117.3 s	
2'''	155.6 s		163.5 s		155.7 s	
3'''	103.2 d	6.425 d 2.4	103.8 d	6.440 d 2.4	103.8 d	6.521 s
4'''	158.3 s		°158.3 s		154.1 s	
5'''	107.3 d	6.423 dd 9.0;2.4	108.0 d	6.505 dd 8.4;2.4	109.2 s	
6'''	128.9 d	7.430 d 9.0	131.7 d	7.390 d 8.4	128.6 d	7.768 s
C'	81.0 d	5.491 d 6.0	42.8 d	4.884 d 11.8	81.8 d	5.287 d 6.0
α'	51.9 d	3.191 dddd 13.1;6.0;4.4;3.7	52.8 d	4.708 dd 11.8;3.1	52.4 d	3.205 m
β'	36.5 t	2.414 dd -13.4;4.4 2.241 dd -13.4;13.1	90.7 d	5.581 d 3.1	36.7 t	2.479 dd 13.6;4.1 2.147 dd 13.6;13.6
1''	131.4 s		128.6 s		134.1 s	
2''	130.9 d	6.758 m	128.3 d	7.086 m	131.2 d	6.782 m
3''	116.1 d	6.563 m	116.3 d	6.791 m	116.3 d	6.598 m
4''	156.4 s		°157.9 s		*156.6 s	
5''	116.1 d	6.563 m	116.3 d	6.791 m	116.3 d	6.598 m
6''	130.9 d	6.758 m	128.3 d	7.086 m	131.2 d	6.782 m
OH		12.860 s		13.612 s		12.853 s

(* , ° , ° may be reversed; δ ppm; J Hz; m: multiplicity).

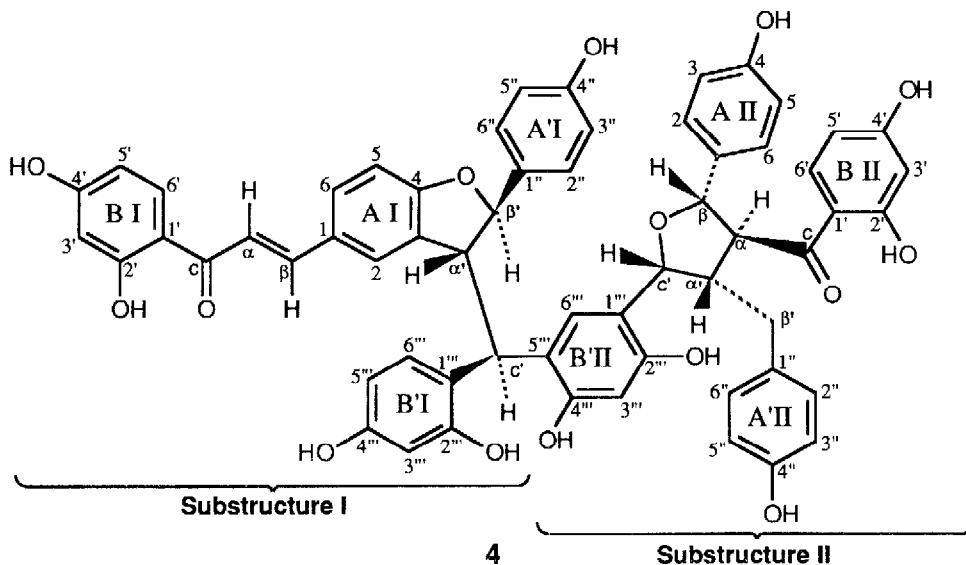
methyl ether derivative **3** (^1H NMR, $\text{Me}_2\text{CO}-d_6$, $\delta_{\text{CH}_3\text{O}}$: 3.859, 3.809, 3.806, 3.802, 3.616, and 3.336 ppm), confirming that all the six hydroxyl groups are phenolic.

The ^{13}C NMR spectrum of **1** had signals of all 30 carbon atoms in the molecule, among which 1 carbonyl, 24 aromatic, and 5 aliphatic (2 $>\text{CHO}$, 2 $>\text{CH}$ and 1 CH_2) carbons (Table 1). Its ^1H NMR spectra (1D and 2D COSY) revealed four benzene rings (a *p*-hydroxyphenyl, a 2,4-dihydroxyphenyl, a *p*-hydroxybenzyl and a 2,4-dihydroxybenzoyl groups) and a 2,3,4,5-tetrasubstituted tetrahydrofuran ring. A singlet signal at 12.860 ppm collapsing slowly with D_2O addition was assigned to a strongly chelated hydroxyl group. The ^{13}C NMR INADEQUATE spectrum of **1** showed connections between these rings: the benzoyl carbonyl carbon atom ($\delta_{\text{C}} = 205.5$ ppm) was linked to both the quaternary carbon, C-1' ($\delta_{\text{C}} 113.5$) and the methine carbon C- α ($\delta_{\text{C}} 59.6$) while the methylene carbon C- β' ($\delta_{\text{C}} 36.5$) had correlations with the quaternary carbon C-1'' ($\delta_{\text{C}} 131.4$) and the methine

carbon C- α' (δ_C 51.9). Finally the methine carbons C- β (δ_C 84.0) and C-c' (δ_C 81.0) were connected to C-1 (δ_C 133.3) and C-1''' (δ_C 117.3) respectively leading to structure 1 for isombamichalcone.

The relative configuration of the tetrahydrofuran ring substituents of 1 was determined from ^1H NOE measurements as *trans-trans-cis*, since low values were observed between the vicinal protons α -H and α' -H (0.5%), α -H and β -H (2.9%) and higher values between c'-H and both α' -H (8.2%) and β -H (11.5%). This compound is thus an isomer of mbamichalcone in which the substituents have an all *trans* relative disposition [3].

Lophirochalcone, 4, amorphous brown solid, $[\alpha]_D^{25}$: +181° (c 0.6, Me₂CO) gave $[\text{M-H}]^-$ ion at m/z 1007 in FAB(-) MS consistent with molecular formula C₆₀H₄₈O₁₅. IR absorptions were indicative of hydroxyl (3409 cm⁻¹), conjugated and chelated C=O, (1626 cm⁻¹), and aromatic ring (1515 cm⁻¹). The presence of eleven hydroxyl groups in 4 was evident from the fact that its complete acetylation (Ac₂O/Py) gave an undeca-acetate (C₈₂H₇₀O₂₆), the ^1H NMR of which showed eleven methyl signals from AcO at 2.310, 2.302, 2.274, 2.264, 2.243, 2.195, 2.171, 2.164, 2.152, 2.048, and 2.037 ppm.



The spin systems identified from ^{13}C and ^1H NMR (1D and 2D COSY) spectra of 4 (Table 1) were eight benzene rings (one 1,2,4,5-tetrasubstituted, three 1,2,4-tri-substituted, three 1,4-disubstituted phenyls and a dihydrobenzofuran ring), a tetra-substituted tetrahydrofuran ring, a methine and a methylene groups, a disubstituted *trans* double bond and two carbonyls. Singlet signals at δ_H : 12.853 ppm (1H) and 13.612 ppm (1H) were assigned to two hydroxyl groups *peri* to carbonyl groups.

Chemical shifts and coupling constants from ^1H NMR spectrum of **4** were very similar to those for isombamichalcone (**1**) and lophirone C [2]. Examination of the long range ^1H - ^1H COSY spectrum of **4** established two substructures (I and II): I is similar to lophirone C, but has the carbonyl carbon c' reduced to a >CH- group, while II is basically isombamichalcone, except that it has only two protons (3'' and -6'')-H on the B'-II ring. Cross peaks observed between 6'''(ring B'-II) -H and the methine proton (c',I)-H at $\delta_{\text{H}} = 4.884$ show that the interflavonoid bond is between the carbons C-c' from substructure I and C-5''' from substructure II.

NOE difference experiments and the ^1H NOESY spectrum of lophirochalcone **4** allowed its relative stereochemistry to be defined. In substructure I, a weak NOE effect between β' -H and α' -H (2.9%) indicated their *trans* disposition on the dihydrofuran ring. No NOE was observed between α' -H and C'-H confirming that they are in a favoured *anti* conformation, in agreement with the large value (11.8 Hz) of their mutual coupling constant. But strong NOEs were measured between C'-H and β' -H (7.8%) which are nearby in space and also between α' (I)-H and each of the three protons (2'',6'') (A'-I)-H (8.8%), 6'''(B'-I)-H (7.4%) and 6'''(B'-II)-H (6.8%). A NOE effect (1.6%) observed between 6'''(B'-II)-H and β' (I)-H implies that they are *gauche* and thus allows assignment of ring B'-I and B'-II respective positions. The relative stereochemistry of the tetrahydrofuran ring (substructure II) was deduced from the strong NOEs observed between c'(II)-H and both α' (II)-H (10.0%) and β' (II)-H (8.3%) that are all *cis*, whereas α' (II)-H was placed *trans* because it gave a weak NOE (1.2%) with β' (II)-H. Thus the relative stereochemistry of substructures I and II is the same as in lophirone C and **1** respectively. Significant NOEs observed between aromatic protons (2'',6'') (A'-I)-H and the methylene protons β' (0.9 and 1.6%) of substructure II imply their spatial proximity and can be only explained by the relative stereochemistry of substructure I as compared to that of substructure II depicted in **4**. These results were confirmed by analysis of the ^1H NOESY spectrum which showed in addition strong cross peaks between (2,6)-H of ring (A-II) and (2'',6'')-H of ring (A'-I); this data also favoured the relative stereochemistry between substructures I and II indicated in structure **4**.

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